

A study on piperine, active compound of black pepper

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Abstract

Piperine, the most important compound of black pepper, is used for treating several diseases such as breast, colon, rectal, and stomach cancer. In this study, the thermodynamical properties of this molecule in blood were obtained by using DFT and HF at the level of B3LYP/6-31+g(d,p). The results of calculations indicate that Piperine is less stable. So, it can react easily with radicalic forms and decrease their harmful effects and damage to the cell structures. It can dissolve in blood and reach every tissue of the body because of the high polarity.

Key words: Piperine, DFT and HF

Abbreviations

DFT	density functional theory
HF	hartree-fock
HOMO	highest occupied molecular orbital
LUMO	lowest unoccupied molecular orbital
FDA	the food and drug administration
CYP3A4	Cytochrome P450 3A4

1.Introduction

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Black pepper is mainly used for meals among spices. It is worthy for its several bitter quality based on the alkaloid structure of piperine. It is used not only in cooking but also for a lot of aims for example, medicinal, preservative, and perfumery field [1].

Piperine are alkaloid-amide components obtained from Piper species. Several of them have been reported to be cytotoxic activity towards tumor cell lines [4]. Piperine has been reported to display central nervous system depressant, antipyretic, analgesic, and anti-inflammatory activities. Piperine also protects against chemical carcinogens [5]. Moreover, piperine is a potent inhibitor of the mixed function oxygenase system and of P450 isoenzymes [6].

Modern man is confronted with an increasing incidence of cancer and cancer deaths annually. Statistics indicate that men are largely plagued by lung, colon, rectal, and prostate cancer, while women increasingly suffer from breast, colon, rectal, and stomach cancer [2]. The literature indicates that many natural products are available as chemoprotective agents against commonly occurring cancer types [3].

‘Docetaxel is the primarily treatment confirmed by the FDA for prostate cancer. It is metabolized in the liver by hepatic CYP3A4 activity. Piperine, a major plant alkaloid, has been shown to inhibit the CYP3A [4], enzymatic activity in a cell-free system. In a study, It was observed piperine and docetaxel could increase docetaxel’s pharmacokinetic activity in vitro and in vivo. Docetaxel is one of the most commonly used cytotoxic chemotherapeutic agents. Dietary constituents are important agents modifying drug metabolism and transport. Dietary consumption of piperine increases the therapeutic efficacy of docetaxel in a xenograft model without inducing more adverse effects on the treated mice’ [7].

Piperine is an active component of black pepper. It has antimutagenic potential against carcinogens. In vivo, the effect of piperine on serum and tissue glycoprotein levels benzopyrene initiated lung carcinogenesis in Swiss albino mice. Piperine, acts as antioxidant and anticancer agent, was found to be chemoprotective effective especially in lung cancer [9].

Piperine has been effected inhibition of pancreate tumor and breast cancer stem cells. It reduces cancer incidence in lung cancer [10]. It exhibits analgesic and anticonvulsant effects and important for medicinal uses of black pepper in pain and epilepsy [11]. It has treatment effects by activated Caspe-3 prostate cancer cells [12].

As traditional medicines become increasingly popular globally, the significant potential for interaction between traditional medicines and allopathic medicines tends to hog the limelight. Numerous studies have shown that naringin interferes with the activities of transporters and enzymatic proteins in the intestines and, hence, with the absorption and breakdown of certain drugs, resulting in altered blood levels of these drugs [19].

2. Materials and methods

The electronic structures of piperine is studied by first principles methods that contains electronic correlation and spin orbital corrections. DFT and HF, included in. RB3LYP methods were used for geometry optimization. The correction was carried out by means of the 6-31+g(d,p) functional. The thermodynamical values in blood were calculated by using DFT and HF methods. The thermodynamical values in blood were calculated by using DFT and HF method. The correction was carried out by means of the 6-31+g(d,p) functional. These methods and fully optimized geometric structure of the compounds using this method were determined and evaluated [16]. All calculations were performed using the Gaussian 09 program The structure of compound were optimized at the B3LYP/6-31G (d, p) level [17-18].

3. Results and discussions

The values of ΔG (Gibbs free energy), HOMO, LUMO, Δ (HOMO-LUMO) and Dipol Moment related Piperine by using DFT and HF are given in Table 1.

Table 1 The values of ΔG (Gibbs free energy), HOMO, LUMO, Δ (HOMO-LUMO) and Dipol Moment related Piperine by using DFT and HF

Active compound of black pepper	ΔG (Hartree)	HOMO	LUMO	Δ (HOMO-LUMO)	Dipol Moment (Debye)
Piperine DFT (In Blood)	-939.435916	-0.20898	-0.07290	-0.13608	6.3612
Piperine HF (In Blood)	-933.591344	-0.28643	0.06326	-0.22317	5.9750

The structure of Piperine is given in Figure 1.

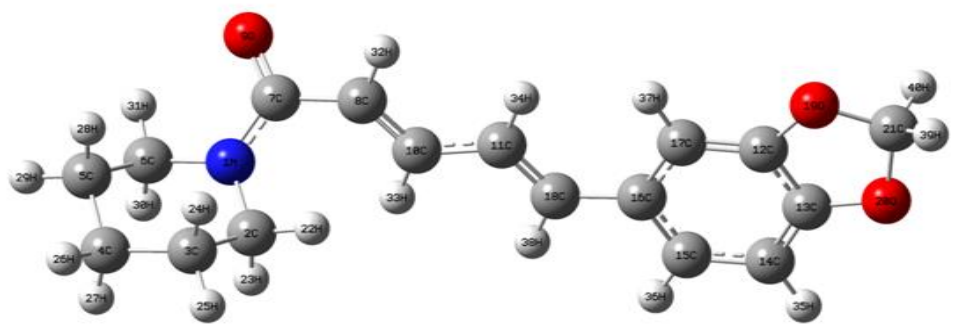


Figure 1 The structure of Piperine

As seen in the table 1: Free energy of Piperine, respectively by DFT and HF, are -939.435916, -933.591344 Hartree. They are rather higher and the difference between HOMO-LUMO is -0.13608 by DFT. Dipol moments of this molecule are respectively 6.3612, 5.9750 debye with DFT and HF.

According to the molecular orbital theory; when it occurs atoms within molecules necessary bond creates atomic orbitals mixed molecule of molecular orbitals leading to formation they approach each other. These orbitals can be considered as one where there is greater probability of electrons in molecules [13]. The gap between HOMO and LUMO indicates the stability of the compound [14].

Ionization potential (I) is the energy to move away one e⁻ from the molecule in gas phase.

$$I = -E_{\text{HOMO}}$$

Ionization potential of Piperine in blood phase by using HF is 0.28643eV. It is higher than from the other in Table 1.

Electron affinity (A) is the energy of a molecule added one e⁻.

$$A = -E_{\text{LUMO}}$$

Electron affinity of Piperine in blood by using DFT is -0.07290 eV. It is higher than from the other in Table 1.

The datas from DFT and HF, we concluded that piperine is less stable. So, it can react easily with radicalic forms and prevent their harmful effects to the cell structures. Calculation of the polarity of the measurement of dipole moment, considering the dipole moment vector of each bond is present in the form of the resultant moment vector [15]. He et al. emphasized that the plant's active compound were isolated with ethanol efficiently [21]. Piperine can dissolves in blood and reaches every tissues of the body because of having high dipol moment. Wang et al. emphasized that the lone pairs of the oxygen atom in Y-shaped oxoanions of the molecule directly interact with the -NH groups of the other compouns [20]. In our study piperine may be interact with the blood compounds in the similar way.

Conclusions

Experimental studies that mentioned above also indicated that positive results of these theoretical remarkably similarity. Therefore, the further experimental studies can be done for this active component of black pepper to treat different types of cancer diseases.

Conflict of Interest

The authors state no conflict of interest.

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